3 <i>-</i> 2	Mixtures	(pages 60-						
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	KEY CONCEPTS							
	Matter that consists of two or more substances mixed together but not chemically combined is called a mixture.	A solution is a type of homoge- neous mixture formed when one substance dissolves in another.						
Voca	bulary Skills: Relating Terms							
Tell how	the terms in each pair are related.							
	ion: mixture							
i. solut	ion. imature							
2. solut	e: solvent							
								
3. hetei	ogeneous mixture: homogeneous mi							
3. hetei								
3. hetei								
	ogeneous mixture: homogeneous mi							
	ogeneous mixture: homogeneous mi							
	ogeneous mixture: homogeneous mi							
4. solub	rogeneous mixture: homogeneous mi							
4. solub	ogeneous mixture: homogeneous mi							
4. solub	rogeneous mixture: homogeneous mi							
4. solub	rogeneous mixture: homogeneous mi							
4. solul 5. alloy	rogeneous mixture: homogeneous mi							
4. solul 5. alloy	ogeneous mixture: homogeneous mi							

7.	colloid: mixture	٠.,				
		,		,		
8.	colloid: solution				•	
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Types of Mixtures: Applying the Main Ideas

Complete the diagram below by writing the following terms in the correct boxes:

solutions

heterogeneous

colloids

Then place under the correct heading each of the mixtures listed below the diagram.



